AN EFFICIENT BENZOQUINONE BASED DIELS-ALDER APPROACH TO PODOCARPANE DITERPENES

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<u>Abstract</u>: A short, convergent synthesis of $(\pm)11,14$ -dimethoxypodocarpa-8,11,13-triene, 11, is described based on successful high pressure and Lewis acid catalyzed Diels-Alder reactions for the construction of the 4,4,10-trimethyloctahydrophenanthrene ring system.

The podocarpane unit, 1, is part of the basic carbocyclic framework of several classes of di-, sester-, and triterpenes.¹ Several methods for the preparation of this tricyclic network are known of which the most popular are successive Robinson annulation processes and biomimetic cation-olefin or cation-arene cyclization reactions.² This note reports a new, direct and highly versatile strategy for the construction of this ring system.

The method focuses on the 4+2 cycloaddition reaction of 1,4-benzoquinone, 2, with 2,6,6trimethyl-1-vinylcyclohexene, 3, which is readily available in two steps from inexpensive β ionone in \geq 70% overall yield.³ Diels-Alder reactions of the sterically hindered diene 3 with dimethyl acetylenedicarboxylate (DMAD)^{3,4} and maleic anhydride⁵ have been reported; however, the reproducibility of the latter reaction has been questioned.^{3b} Methoxy- and pyrrolidinoderivatives of 3, 4 and 5, respectively, react with DMAD to give 4+2 cycloadducts 6,^{3b,6} but attempts to effect Diels-Alder reactions with other derivatives of 3 with various dienophiles have generally failed.³ It was not apparent at the outset of this work that Diels-Alder reactions of 3 with dienophiles more sterically demanding than DMAD would succeed.⁷

We now report that a Diels-Alder reaction of 2 and 3 can be effected under high pressure or by Lewis acid catalysis. Thus, a 1:2.8 mixture of 2 and 3, respectively, in CH_2Cl_2 (0.76 M in 2) subjected to 6 Kbar for 14 days affords, after silica gel chromatography (1:4 ethyl acetate:hexanes), endo-cyclohexendione 7 (32%, R_f 0.35, mp 92-93°C), hydroquinone 8 (16%, R_f 0.14, mp 174-176°Cd) and quinone 9 (14%, R_f 0.57, mp 81-82°C).⁶ Identification of 7 as the endo-cycloadduct is based on its 300 MHz pmr spectrum which shows that H_1 is deshielded by the carbonyl oxygen at C-11 and appears as a ddd at 2.38 ppm. In 7, large coupling constants to H_1 of 12 and 12 Hz (J_{gem} and J_{trans} -diaxial) establish that H_1 occupies an axial position. Similar deshielding of C-1 protons in C-11 oxygenated diterpenes has been noted.⁹ Dreiding models of the exo-4+2-cycloaddition adduct of 2 and 3 suggest that the C-11 oxygen would be too far from the C-1 axial hydrogen to have a noticeable effect on its chemical shift. A <u>cis</u> BC ring fusion is indicated by a J_{He}- He = 5 Hz. Hydroquinone 8 apparently results from SiO_2 catalyzed enolization of 7. In a separate experiment, 7 cleanly rearranges to 8 (84% isolated yield) upon treatment with a catalytic amount of aqueous HCl in THF. The source of quinone 9 may involve either a) a high pressure accelerated rearrangement of 7 to 8 which is then oxidized to 9 by 2, or b) a high pressure promoted oxidation of 7 directly to 9 by 2. In control experiments, addition of 1,4-benzoquinone to a solution of 8 in benzene effects complete conversion (tlc, pmr) to 9 with the formation of hydroquinone in ≤ 1 hr; no 9 is present in a CH_2Cl_2 solution of 7 and 2 open to the atmosphere for 4 hr.¹⁰ Hydroquinone 8 is only slowly oxidized by oxygen and can be chromatographed and handled open to the atmosphere for more than a day without serious air oxidation, however, noticeable (tlc) amounts of 9 are present after 48 hr in solution. In addition, 9 is reduced to 8 in 97% yield with zinc dust in HOAc.

A more effective preparation of 7 is accomplished by a Lewis acid catalyzed cycloaddition reaction of 2 and 3. Addition of diene 3 (1 equiv) to a solution of $SnCl_4$ (1 equiv) and 1,4-benzoquinone, 2 (3.4 equiv), in CH_2Cl_2 at -40°C and stirring for 3 hr followed by neutral quench (pH=7 buffer), extraction and chromatography on silica gel gives 7 in 69% yield as the only 1:1 cycloadduct. A 4:1 mixture of 7 and 9 is obtained in 69% yield with $BF_3 \cdot Et_20$ as catalyst at 0°C in CH_2Cl_2 with a catalyst:quinone:diene ratio of 2:4:1. Efforts to obtain cycloaddition adducts from 2 and 3 without catalysis or high pressure produce 7, 8, and 9 in $\leq 10\%$ combined yield after 12 hr at 120°C; at higher temperatures (140° or 160°C), extensive degradation occurs and complex reaction mixtures result.

Both 7 and 8 are methylated with $Na0t-Bu/(CH_30)_2S0_2/THF$ to give 10 (mp 103-104°C) in 65% and 78% yield, respectively, and 10 is hydrogenated in HOAc¹¹ (10% Pd/C, 50 psi, 12 hr) to afford 11 (84%, mp 133-134°C) with a <u>trans</u> AB ring juncture which is identical (300 MHz pmr, ms and tlc) to a sample of 11 prepared from natural podocarpic acid.¹² The ¹H nmr of the crude mixture from hydrogenation of 10 reveals the presence of a second minor product (\leq 8% yield) which we have tentatively identified as the <u>cis</u> isomer 12;¹³ the ratio of 11:12 is ~11:1. Thus, the reaction of 2 and 3 to produce 7 and 8 should provide an expedient synthesis of podocarpa-8,11,13-triene derivatives having a <u>trans</u> AB ring fusion which is the most commonly naturally occuring configuration.^{1,28}

The chemistry presently described is notable in that 1) 10 and 11 are useful intermediates for the synthesis of diterpenes possessing ring B and/or C oxygenation,¹⁺ especially at C-11^{9a} where introduction of oxygen substituents at this carbon in previous strategies has been a difficult problem,¹² 2) 11 has also served as an intermediate in the preparation of labdane diterpenes and drimane sesquiterpenes, including the insect antifeedant warburganal, 13, via oxidative cleavage of the C ring,^{12a,15} and 3) the high pressure and Lewis acid promoted Diels-Alder reactions of 3 and 1,4-benzoquinone appear to be stereoselective; we have found no evidence for the formation of a second stereoisomer of 7. We are currently investigating the reactions of 3 with other quinone dienophiles and the application of these reactions to the synthesis of diterpenes.



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